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#### SUMMARY/ SAMMENORAG

Ten core samples and one oil were analysed to characterise and compare the content of hydrocarbons. The oil is a light crude oil of a high maturity, while all the cores contain an oil with a high wax content. The depth interval between 1699.68m and 1903.50m shows the highest hydrocarbon content. The content decreases both above and below this interval. The quantity of hydrocarbons indicates significant oil saturation, especially in the above mentioned interval (Hunt, 1979).

All the analyses indicate that the hydrocarbons were generated from a source rock with a high input of humic material, that is, a type III or possibly a mixed type II/III kerogen. The maturity of the hydrocarbons is high both in the fluid and in the cores. The hydrocarbons have a somewhat lower maturity in the shallowest, least hydrocarbon saturated cores.

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#### 1. INTRODUCTION

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This report presents the results from the analysis of the reservoir hydrocarbons in well 7121/7-1 from Tromsø-flaket.

The samples received for analysis were as follows:

IKU No.	Sample type	Depth (m)
B-8414	Core	1852.25
B-8415	Core	1862.47
B-8416	Core	1881.71
B-8258	Core	1899.68
B-8259	Core	1900.82
B-8260	Core	1901.70
B-8261	Core	1903.50
B-8417	Core	1913.00
B-8262	Core	1926,50
B-8418	Core	1930.25
B-8257	0il, DST No. 2	1866-71

The samples were assigned the IKU job number 05.1749.00 and the analyses were carried out according to the terms of the open contract T.4533, Job No. 3 from Statoil.

The analyses were as follows:

- 1) 0il/Condensate, 1 sample
  - API gravity
  - GC of whole oil
  - C<sub>2</sub>-C<sub>8</sub> hydrocarbons
  - Chromatographic separation into saturates and aromatics
  - Separation of saturated hydrocarbons by urea adduction
  - GC of saturates, branched/cyclic hydrocarbons and aromatics
  - GC-MS of saturates and aromatics
  - $\delta^{13}$ C isotope analysis of whole oil, saturates and aromatics

2) Core samples, 10 samples

Extraction

- Chromatographic separation into saturates and aromatics

- 5 -



- GC of saturates and aromatics (calculation of MPI)
- GC-MS of saturates and aromatics (2 samples)
- $\delta^{13}$ C isotope analysis of extract, saturates and aromatics

The gas chromatograms are presented as Multichrom plots on a printer/plotter system, and have a relatively poor resolution. This is especially obvious in Figure 1, where pristane (Pr) is seen as a shoulder on  $nC_{17}$ . These two peaks are resolved in the original data file and the  $Pr/nC_{17}$  ratio can be calculated from these data.

The predominance of odd over even carbon numbers, seen particularly in sample B-8414, (Figure 4) is real and is not caused by the data system.

Twenty copies of this final report have been made, Statoil has been sent ten, and ten are kept at IKU.

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#### 2. EXPERIMENTAL

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#### 2.1 Extractable Organic Matter

Powdered rock was extracted by flowblending for 3 minutes using dichloromethane (DCM) and methanol (1%) as solvent. The DCM used was of organic geochemical grade and blank analyses showed the occurrence of negligible amounts of contaminating hydrocarbons. Activated copper fillings were used to remove any free sulphur from the samples. After extraction the solvent was removed on a Buchi Rotavapor and the amount of extractable organic matter (EOM) was determined. Asphaltene precipitation was performed according to the procedure given by Statoil.

#### 2.2 Chromatographic separation

The extractable organic matter (EOM) was separated into saturated fraction, aromatic fraction and non hydrocarbon fraction using a MPLC system with hexane as eluant (Radke et al., Anal. Chem., 1980). The various fractions were evaporated on a Buchi Rotavapor and transferred to glass vials and dried in stream of nitrogen.

The same separation procedure was applied to the fractions of oil/condensate samples boiling above  $210^{\circ}$ C.

#### 2.3 Urea adduction

Urea-adduction was performed on the saturated hydrocarbon fraction. The sample containing 5 mg of n-alkanes was dissolved in 2 ml of n-hexane and 1 ml of acetone was added. A saturated solution of urea in methanol (1 ml) was then added dropwise. The solvent was removed  $(N_2)$  and the adduction step repeated twice. The white crystals were rinsed (3x5ml hexane) and the combined extract filtered (cotton wool plug covered with  $Al_2O_3$ ), to afford a non-adduct. GC analyses were performed on the samples after the urea adduction, using the same conditions as for the other GC analyses.

#### 2.4 Gas chromatographic analysis

Whole oil samples were analysed on a HP5730A GC, applying a temperature program from  $-50^{\circ}$ C (2 min.) to  $280^{\circ}$ C at  $4^{\circ}$ C/min. The GC was fitted with

a 15m DB-1 fused silica column. Hydrogen (2.5ml/min.) was used as carrier gas, and 0.02µl was injected in split mode (split ratio 1:10).

The  $C_2-C_8$  hydrocarbon fraction was determined on a Carlo Erba Fractovap GC. The column used was a 30m fused silica capillary column coated with SE-54. The temperature program applied was  $50^{\circ}C$  (2min.) to  $180^{\circ}C$  at  $4^{\circ}C/min$ .

The saturated, the branched/cyclic and the aromatic hydrocarbon fractions were each diluted with n-hexane and analysed on a HP 5730A. The GC is equipped with a 15m DB-1 fused silica column and hydrogen (ca. 2.5 ml/min.) is used as carrier gas. Injections are performed in split mode (split ratio 1:10). The temperature program applied is  $80^{\circ}C$  (2 min.) to  $280^{\circ}C$  at  $4^{\circ}C/min$ .

The data processing for all the GC analyses was performed on a VG Multichrom lab data system.

#### 2.5 Gas chromatography - mass spectrometry (GC-MS)

GC-MS analyses were performed on a VG Micromass 70-70H GC-MS-DS system. The Varian Series 3700 GC was fitted with a fused silica OV-1 capillary column (30m x 0.3mm i.d.). Helium (0.7kg/cm<sup>2</sup>) was used as carrier gas and the injections were performed in split mode ( $1.5\mu$ ], split ratio 1:15). The GC oven was programmed from 70<sup>o</sup>C to 280<sup>o</sup>C at 4<sup>o</sup>C/min. after an initial isothermal period of 2 minutes.

The saturated hydrocarbons were analysed in multiple ion mode (MID) at a scan cycle time of approximately 2 secs. Full data collection was applied for the aromatic hydrocarbons at a scan time of 1 sec/decade. The mass spectrometer operated at 70eV electron energy and an ion source temperature of  $200^{\circ}$ C. Data acquisition was done by VG data systems.

Peak identification was performed applying knowledge of elution patterns in certain mass chromatograms. Calculation of peak ratios was done from peak height in the appropriate mass chromatograms.

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The  $\delta^{13}$ C isotope analysis was performed by mass spectrometry at Institute for Energy Technology (IFE) in Oslo according to their method. Their reference value for the standard NBS-22 is -29.8.

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#### 3. RESULTS AND DISCUSSION

#### 3.1 API gravity

The API gravity of the oil (Table 1) was determined to be  $52.4^{\circ}$  API. This indicates that the oil is a light crude oil, with a high content of gasoline range hydrocarbons.

#### 3.2 GC of whole oil

The gas chromatogram of the whole oil is presented in Figure 1. The normal alkanes in the range  $nC_{6}-nC_{15}$  are abundant. Cycloalkanes and aromatics such as benzene and toluene are also prominent. This suggests either a source rock with a high aromatic content, i.e. a terrestrial humic type, or possibly that biodegradation of the oil may have occurred.

## 3.3 C<sub>2</sub>-C<sub>8</sub> hydrocarbons

The quantitative data and the gas chromatogram from the  $C_2-C_8$  hydrocarbon analysis are presented in Table 2 and Figure 2, respectively. The gas chromatogram of the light hydrocarbons presents similar results to those obtained from gas chromatography of the whole oil and shows abundant cycloalkanes and aromatics in addition to the n-alkanes. The ratio between branched and normal alkanes is less than unity both for  $iC_4/nC_4$  (0.33) and  $iC_5/nC_5$  (0.67), suggesting that no heavy biodegradation has taken place. This would indicate that the high aromatic content is due to a type III or II/III source rock. The high aromatic content also indicates a highly mature condensate.

#### 3.4 Extraction and separation

Ten core samples were extracted before chromatographic separation of the extractable organic matter. The light components of the oil, boiling below 210<sup>0</sup>C, were evaporated before chromatographic separation. The weights and relative amounts of the various chromatographic fractions are given in Tables 3-6.

All ten cores contain high amounts of extractable organic matter. The depth interval between 1899.68 to 1903.50m has the highest concentra-



tion of extractable organic matter. Based on the crossplot of organic carbon and ppm  $C_{15}^+$  hydrocarbons, most of the ten cores fall in the zone for reservoir rockss or very close to the borderline between this zone and the nonreservoir zone (Figure 3). The samples in the interval mentioned above, 1899.68-1903.50m, fall in the upper part of the reservoir zone. The two deepest cores at 1926.50-1930.25m, also have values in the reservoir zone. The light crude oil was taken approximately 30m above the depth interval of the richest cores.

The oil contains approximately 14% saturated and 6% aromatic hydrocarbons which gives a SAT/ARO ratio equal 2.4 and a hydrocarbons/nonhydrocarbons ratio of 0.2. The SAT/ARO ratios of nine of the cores are quite similar to that of the oil (1.6-2.7). Core B-8417 at 1913.00m, which also contains a poor amount of hydrocarbons, shows a lower SAT/ARO ratio of 1.3. The hydrocarbons/non-hydrocarbons ratios for four of the cores (B-8414, B-8415, B-8416 and B-8262) are similar to that of the oil. The other cores, except for B-8417, have a higher hydrocarbon content i.e. hydrocarbon/non-hydrocarbons ratios of 0.5-0.75. This high relative hydrocarbon content occurs in the interval with richest hydrocarbon content.

### 3.5 Gas chromatography of $C_{15}^+$ saturated hydrocarbons

Gas chromatograms of the saturated hydrocarbon fractions are presented in Figure 4 and data from the gas chromatograms are given in Table 7.

The gas chromatogram of the oil shows a smooth, front-biased n-alkane distribution with a maximum at  $nC_{14}$  and poor development of the heavy molecular weight hydrocarbons.

The gas chromatograms of the ten cores are all very similar, and show a quite different pattern compared to that of the oil. A unimodal n-alkane distribution with a maximum at approximately  $nC_{22}$  is seen in the core extracts' gas chromatograms. Five of the cores show a low relative amount of internal standard which is due to the presence of a higher abundance of migrated hydrocarbons in these samples. This is supported by the high hydrocarbon/non-hydrocarbon ratios of the same samples. The distribution of n-alkanes suggests a waxy oil that has been generated from a source with high input of terrestrial humic material.

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The pristane/nC<sub>17</sub> ratios of the ten cores and the oil are all similar, while the pristane/phytane ratios show differences. The pristane/phytane ratios of the cores range from 1.4 to 1.7, (excluding the shallowest core which has a ratio of 0.9), while the pristane/phytane ratio of the oil has a higher value of 2.4. This would also indicate a source rock which has had a high input of terrestrial material, possibly a mixed type II/III kerogen. The differences between the oil and the core extracts may be due simply to the generally high content of lower molecular weight compounds in the oil.

The CPI values of the core extracts are low (0.9-1.1) and suggest that the samples are mature, with the possible exception of the shallowest core (B-8414) which has had an input of less mature hydrocarbons. Two explanations can be suggested for the origin of the hydrocarbons in the oil and the cores. One is that they have all been generated from the same source and the light oil has been separated from the rest through migration. This is a likely explanation fince the light oil is reservoired at a shallower depth than the core intervals which have the highest quantity of hydrocarbons. The second explanation is that different type of source rocks have generated the oil and the core hydrocarbons.

### 3.6 Gas chromatography of $C_{15}$ + branched/cyclic hydrocarbons

The saturated hydrocarbon fraction of the oil, B-8257 was separated by urea adduction. The gas chromatogram of the branched/cyclic hydrocarbons is presented in Figure 5. The pristane/phytane ratio of 2.4 is the same as was obtained from the gas chromatogram of the saturated hydrocarbon fraction. The usual  $C_{15}$ - $C_{20}$  isoprenoids are the main components and polycyclic compounds do not appear to be abundant. This indicates an oil of high maturity that has been sourced from terrestrial matter.

### 3.7 Gas chromatography of $C_{15}^{+}$ aromatic hydrocarbons

Table 8 gives the MNR, DMNR, MPI 1 and MPI 2 ratios calculated from the gas chromatograms and Figure 6 presents the gas chromatograms of the aromatic hydrocarbons.

The methyl- and dimethyl-naphthalenes are most prominent in the gas chromatogram of the oil, while phenanthrene and it's alkylated homologs 142/p/jbl/11

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#### 3.8 GC-MS analysis of saturated terpanes and steranes

Mass chromatograms representing terpanes (m/z 191) and steranes (m/z 217 and 218) are shown in Figure 7 and molecular ratios calculated from the chromatograms are given in Tables 9 and 10.

Only two of the core sample extracts were analysed by GC-MS, together with the oil sample. The general distribution of terpanes is dominated by the ubiquitous  $17\alpha$ ,  $21\beta$ -hopanes in the two core extracts. The oil has a high relative abundance of tricyclic terpanes (P-T in m/z 191) in good agreement with a light crude/condensate type oil. The high abundance of the peak A is probably due to a contaminant. Squalene has been identified in other samples with low amounts of terpanes and has been seen to coelute with one of the trisnorhopanes, Ts (Peak A). Apart from the higher input of lower molecular weight terpanes in the oil, the distribution of terpanes in the oil is most similar to that of the extract from the deepest of the two cores at 1903.50m (B-8261). The other core extract seems to contain hydrocarbons of lower maturity. None of the samples contain the 28,30-bisnorhopane that is seen in many samples from the North Sea.

A lower maturity for the uppermost core sample at 1862.47m, is also suggested by the distribution of both the %pp and %20S steranes. The sterane chromatogram of oil sample suggests the presence of the same

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contaminant as was seen in the m/z 191 trace, i.e. a squalene peak coeluting with peak j in the m/z 217 chromatogram. Apart from this peak the sterane trace of the oils is dominated by low molecular weight steranes (u, v in m/z 217) and rearranged steranes. This is in good agreement with the other data for the oil and indicates that the oil has been generated from a mature source rock. The distribution of 14 $\beta$ ,17 $\beta$ -steranes (m/z 218) shows that both core sample extracts contain higher relative amounts of C<sub>29</sub> steranes, suggesting a source rock with high input of terrestrial organic matter. A comparison of the oil sample with the core extracts is complicated by the higher amount of low molecular weight compounds in the oil. Based on these data it is not possible to say if one source is responsible for the hydrocarbons in the cores and for the oil. There is, however, no information in the GC-MS data that excludes one common source.

#### 3.9 GC-MS analysis of aromatic hydrocarbons

Mass chromatograms are presented in Figure 8 and certain aromatic ratios are given in Table 11.

As was already discussed on the aromatic hydrocarbon GC's, the oil contains abundant naphthalenes and hardly any phenanthrenes or other higher molecular weight compounds. Both core extracts have a higher input of higher molecular weight aromatics. The main difference in the mass chromatograms are the shorter benzene alkyl chains that are seen in the oil. The internal distribution of naphthalenes, phenanthrenes and dibenzothiophenes is very similar in all three samples, while only the deepest core showed any content of aromatic steranes.

## 3.10 s<sup>13</sup>C isotope analysis

The results from  $\delta^{13}$ C isotope analysis of extracts and hydrocarbon fractions are given in Table 12.

The maximum variations within the individual fractions are less than 1 unit for the ten core samples. The isotope values for the oil are for all three fractions slightly above the intervals seen for the same fractions of the core samples. This suggests that the source of the oil is isotopically fairly similar to the source of the core hydrocarbons. The low  $\delta^{13}$ C values support the suggestion that a terrestrial source has generated the hydrocarbons.



#### 4. CONCLUSION

Ten core samples and one oil were analysed to characterise and compare the content of hydrocarbons. The oil is a light crude oil of high maturity, while all the cores contain an oil with a high wax content. The depth interval between 1699.68m and 1903.50m shows the highest hydrocarbon content. The hydrocarbon content decreases both above and below this interval. The quantity of hydrocarbons indicates significant oil saturation, especially in the above mentioned interval (Hunt, 1979).

All the analyses indicate that the hydrocarbons were generated from a source rock with a high input of humic material, that is, a type III or possibly a mixed type II/III kerogen. The maturity of the hydrocarbons is high both in the fluid and in the cores. The hydrocarbons have a somewhat lower maturity in the shallowest, least hydrocarbon saturated cores.

It is difficult to say whether the same source rock is responsible for the oil and the core extracts, but none of the data excludes a common source.

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#### APPENDIX

### Molecular ratios from terpane and sterane mass chromatograms applied as maturity and source characteristic parameters

Geochemical fossils or biological marker components are characteristic of the type of organic matter present at the time the sediments were deposited. The biological isomers of these components undergo changes due to increased maturity in particular, but also to a certain degree caused by migration and weathering processes.

#### Source characteristic parameters

In the m/z 191 mass chromatograms which represent the terpanes, the hopanes and moretanes are the major components in most extracts and oils. Of the hopanes the  $C_{27}$  and  $C_{29}$ - $C_{35}$  homologs are ubiquitous, while the  $C_{28}$  bisnorhopane is believed to be typical of certain types of source rocks. This is also the case for the component, probably gammacerane, sometimes seen to coelute with the 22S isomer of the  $C_{31}$   $17_{\alpha}(H)$ -hopanes (H). In the sterane mass chromatograms, m/z 217 and m/z 218, the molecular weight distribution of the  $C_{27}$ - $C_{29}$  regular steranes is believed to be representative of the original input of organic matter. The highest molecular weight compounds, the  $C_{29}$  steranes, represent organic matter of terrestrial origin, while the lower molecular weight analogs originate from more marine type environments.

#### Maturity dependant parameters

The biological isomers of the hopanes, the  $17_{\beta}(H)$ ,  $21_{\beta}(H)$ -hopanes, undergo structural changes during the maturation process. The isomerisation reactions are thought to be produced via the  $17_{\beta}(H)$ ,  $21_{\alpha}(H)$ -hopanes (moretanes) to the most stable  $17_{\alpha}(H)$ ,  $21_{\beta}(H)$ -hopanes. At equilibrium 100% of the  $17_{\alpha}(H)$ -hopanes are seen. The ratio  $\alpha\beta/\alpha\beta+\beta\alpha$  is used to describe this reaction. In the extended hopanes ( $\geq C_{31}$ ), the thermally stable S configurations at C-22 become increasingly more abundant as compared to the biologically preferred R configurations at increased maturity level. The equilibrium ratio is approximately 60% of the 22S configuration. Another ratio that is known to change with maturity is the Tm/Ts (Seifert et al., 1978) of the C<sub>27</sub> hopanes. The maturable  $18_{\alpha}(H)$ -trisnorneohopane (Tm) is reduced in intensity relative to the more stable  $17_{\alpha}(H)$ -trisnorneohopane (Ts), causing the Tm/Ts to decrease at increased





maturity. This ratio is also believed to be source dependant, and this should be born in mind when applying the ratio for maturity comparison. The amount of tricyclic terpanes is also to a certain extent seen to be maturity dependant.

Two isomerisation reactions taking place in the steranes are most commonly applied for maturity assignments from the m/z 217 mass chromatograms. The biologically preferred  $14\alpha(H)$ ,  $17\alpha(H)$ -isomers of the regular steranes is transformed to the thermally stable  $14\beta(H)$ ,  $17\beta(H)$ -steranes, the % $\beta\beta$ approaching 75% at equilibrium. An equilibrium concentration of 50% is seen of the stable S configuration at C-20 as opposed to the 100% of the biological 20R epimer (Mackenzie et al., 1980). The abundance of rearranged steranes increased with increasingly maturity.

One of the reactions taking place at an early stage of diagenesis is the aromatisation of steranes, leading to the formation of mono- and tri-aromatic analogs. This process is measured as the abundance of triaromatic relative to mono-aromatic compounds (% tri/tri + mono) in the m/z 231 and 253 mass chromatograms, respectively. In addition the degree of side chain cracking, as  $%C_{20}/C_{26}$ , 27 and  $%C_{21}/C_{28,29}$  respectively, is applied. These cracking processes are also taking place during early diagenesis, and are used for maturity assignment together with the previously mentioned ratios.

Migration and weathering

The effect on the geochemical fossils of migration and weathering, is less apparent than the maturity induced changes. Migration is believed to cause an increase in the relative amounts of rearranged and  $14\beta(H)$ ,  $17\beta(H)$  regular steranes (Seifert and Moldowan, 1978, 1981). Severe biological alteration leads to the formation of desmethyl-hopanes (Seifert and Moldowan, 1979).

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## Table 1. API gravity.

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IKU no.	Code	Depth (m)	Density	API gravity
B-8257	DST no.2	1866-71	0.7686	52.4 <sup>0</sup>



1. 19 19 19 19

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<u>Table 2.</u> Light hydrocarbons C<sub>2</sub>-C<sub>8</sub>. B-8257

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q	of	+0	ta'	1	oil
<i>R</i> 0	Vi -	- 60	ιa	Ł	<b>VI</b> 7

 $\{\eta_{i,1}\}_{i\in \mathbb{N}}$ 

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C <sub>2</sub>	
3       MC_ (iC_)	0.3
nC.	0.9
MC. (iC_)	1.6
nC_	2.4
CvC_+2.3DMC_+2MC_	2.4
3MC-	1.2
nC <sub>c</sub>	3.7
MCvC~	2.2
Benzene	1.9
CvCc	2.6
2MC -	1.5
2,3DMC <sub>c</sub>	0.5
3MC <sub>c</sub>	1.9
o DMCyC <sub>E</sub>	2.1
nC <sub>7</sub>	4.4
MCyCe	6.5
Toluene	4.9
2MC <sub>7</sub>	1.8
3MC <sub>7</sub>	1.4
DMCyCe	1.3
nC <sub>e</sub>	4.2
(m+p)-xylene	2.3
o-xylene	2.0

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Abbreviations

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C <sub>2</sub> C <sub>2</sub>	ethane propane
$MC_{3}(iC_{4})$	methylpropane
nCa	butane
$MC_{4}^{T}$ (iC <sub>5</sub> )	methylbutane
nC <sub>5</sub>	pentane
CyC <sub>5</sub> +2,3DMC <sub>4</sub> +2MC <sub>5</sub>	cyclopentane+2,3dimethylbutane+2methylpentane
3MC <sub>5</sub>	3-methylpentane
nC <sub>6</sub>	hexane
MCyC5	methylcyclopentane
сус <sub>6</sub>	cyclohexane
2MC6	2-methylhexane
2,3DMC5	2,3-dimethylpentane
3MC <sub>6</sub>	3-methylhexane
DMCyC5	${\tt lcis3dimethylcyclopentane+2tr2dimethylcyclopentane}$
nC <sub>7</sub>	heptane
MCyC6	methylcyclohexane
2MC7	2-methylheptane
3MC7	3-methylheptane
DMCyC	dimethylcyclohexane
nC <sub>8</sub>	octane

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TABLE : 3.

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#### CONCENTRATION OF EOM AND CHROMATOGRAPHIC FRACTIONS

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***	******	# = 1	13x×xxp=3=	72223 <b>2</b> 22		Azz=#233		*=====	*******	1=26====	*******	= =
I				: Rock	1		3		:	Non	:	1
I	IKU-No	5	DEPTH	: Extr.	: EOM	: Sat.	28	Aro.	: HC	: HC	: TOC	I
ĩ		:		8	1	1	\$		1		2	I
I		ž	(m)	: (g)	: (mg)	; (mg)	ŧ,	(mg)	: (mg)	: (mg)	: (%)	I
I		ť		:	5	: -	ŧ	-	:		1	I
I==	********	** * -		*******	******	*******	5 22 ;	*****	*******	******	******	= I
I		2		н Н	R 0	:	1		5	5	d.	I
I	8 8414	ŧ	1852.25	: 28.3	: 67.5	: 8.0	ĩ	4.0	: 12.0	55.5	: 0.41	I
I		2		;	:	9	*		;	ł	1	1
I	8 8415	3	1862.47	: 36.0	: 92.8	: 9.1	4	4.0	: 13.1	: 79,7	: 0.19	1
I				ä	1	<b>*</b>	ž		5	t.	:	1
1	B 8416	Ę	1881.71	: 27.2	61.4	: 7.2	8	2.8	: 10.0	51.4	: 0.15	1
Ι		ų,		1	;	•	1		3	•	:	I
I	8 8258	1	1877,68	: 26.9	: 172.5	: 40.6	;	15.7	: 56.3	: 116.2	: 0.41	I
I		5		*	1	•	2		<b>;</b>		:	I
I	8 8259	3	1900.82	: 28.3	: 381.6	87.6	;	48.5	: 136.1	: 245.5	: 0.68	I
I		;			;	1	1		" •		:	1
I	B 8260	2	1901.70	: 21.8	: 299.9	: 87.8	1	32.0	119.9	180.0	: 8.62	1
X.		2		:	1	*	:				*	1
ĩ	8 8261	7	1903.50	: 28.9	: 319.4	\$ 80.2	1	37.8	: 118.8	201.4	1 U.45	1
1		2	4.004.00		1		;		•		1	i
i v	8 6417	ļ.	1913	: ລູບ.ລ	: 125.1	: 6.Z	1	÷.∀	: 11.Z	114.9	: U.18	1
1	5.00/0	-	100/ 50	. de o	. 70 0							1
L T	8 9494	5	1728.30	1 20.Q	: / <i>L</i> . <i>L</i>	10.7	1	4.7	: 10.0	1 30.4	: U.I/	1
1 T -	0 0/10	2	1070 95	. 77 /	. 101 E		:	42.7	₹ - /*7 L		: . n/=	1
7 1	0 0410		1730.23	: 33.4	. 101-0	\$ 27.2 -	2	14.4	; 43.0 . ·		: U.4⊒	T.
ц Т	8 8787	i	1922-71	. 074.7*	701 1**	475	i	17 L	• • 401 1	7/15	: 	4 7
ъ Т	0 020/ mil	ž	1000-71	4 7139+/ *	. 901'O	లో చులు ల	•	1/*0	r Ovil	፥ ድጥደነት የ	» ա.ա.ա. «	T
1 7 7 7	4.4W 	; ≃∞-	2"''GU Tabababar	*	*	•	• • • • •		**********	) *******	*	

DATE : 7 - 1 - 85.

\* Amount of crude oil (mg).

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\*\* Amount of oil boiling above 210<sup>0</sup>C.

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## TABLE : 4.



#### WEIGHT OF EOM AND CHROMATOGRAPHIC FRACTIONS

### (Weight ppm OF rock)

	********	***********		:= = =	******	z 3 a	*285475	=	*******	±⇒≏		≘₽
I		f	1	Ţ.		5		;		1	Non	ĭ
I	IKU-No	: DEPTH	i Eom		Sat.	ŧ	Aro.	;	HC	;	HC ·	1
Ţ		;	2	:		5		1		ĩ		I
I		t (m)	:	:		:		1		R.		Ĭ
Ĩ≈≂	*********		********	***	*****	* * *		==:	*******	***	*******	≓I
1		*	1	2		<b>*</b>		:		:		I
ĩ	B 8414	: 1852.25	: 2384	#	284	5	140	;	424	8	1960	Ĩ
I		:	1	÷		ā		\$		4		I
I	B 8415	: 1862.47	: 2578	1	253	:	110	3	363	\$	2214	1
I		:	:	ţ.		2		ŧ,		1		1
I	B 8416	: 1881.71	: 2255	:	264	:	101	ĩ	366	:	1887	I
I		1	1	1		ĩ	* * *	:		4		I
1	8 8258	: 1899.68	6420	1	1509	1	282	2	2095	1	4325	i
i t	D 0050			:	*****	:	4774/	F	1000	-	8/ <b>9</b> 2	Ţ
1	8 8209	1700.62	1 13484		3070	5	1/14		4809	ž	66/3	7
L T	0 90/0	. 1001 70		1	1000	:	11.70	:	s ( 00		0 M E 0	1
Ť	B 020U	1401*10	- - -	1	4UZ7	1	14/0	1	2477	1	Ø238	Ŧ
1 7	B 90/1	: . 4007 80	\$ • • • • • • • • • • • • • • • • • • •	2		1	*7*0	2	1027	1	6076	+
л Т	D D201	: 1703.30	- 11065	:	4///	2	7910	2	4U07	•	07/10	1 7
ц т	D 94+7	. +017	. <u>രജന</u> ന	ă	1.77	i	<b>60</b>	÷.	994	-	0078	Ţ
т	D 041/	- 1713	i	ē v	144		70	ų v	الا يتك يك	i 1	22/0	r T
1 T	D 2747	: . 1071 SD	. 7200	÷.	477	ĩ	101	2	616	ā L	7125	Ţ
Ţ	D 0202	; 1720,30 ,	* 2000 *		420	•	171		Q1-7	-	71Q-1	4 7
л Т	D 8610	* 1070 75	• • "30-74		277		431	,	1 7/17	*	1777	Т
1 7	D 0710	* 1700*****	• ••••••		0/2	:	- e e a a	*	* 000	•	17.00	r
Ŧ	B 8257	• • 1844-71	4 #					•		•		1 7
+ 7	ມັບເມາ ດ.1	• ngt-0	*	."		•		;		*		Ť
			, 238058255		******	, ≍⇒≭						

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### TABLE : 5.



CONCENTRATION OF EOM AND CHROMATOGRAPHIC FRACTIONS

I.

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### (mg/g TOC)

<b>3 3</b> 8		1 <i>2 3</i>		=#	*****	****	******	= = =		***	******	= = :		¥ #
I		1		1		3		8		:		5	Non	I
I	IKU-No	,	DEPTH	;	EOM	1	Sat.	2	Aro.	ŝ	HÇ	1	HC	I
1		1		1		ž		1		ž		1		I
I		\$	{m}	۰ <b>#</b>		1		÷		1		2		I
]==		: # #		= =	****	2223		222	******	====			********	= Į
I		ť				4		1		1		3		Ĩ
I	B 8414	ä	1852.25	;	581,5	3	67.3	5	34.1		103.4	2	478.2	I
Ĩ		2		1		3		1		â		e.		1
1	8 8415	3	1862.4/	Ĩ	1356.7	1	133.3	1	57.9	1	141.2	1	1165.5	i
i v	***	-		4	4807 6	1	****	1	, <del>,</del> ,	1		1		1
1	p 0410	3	1001-/1	4	1000.2	1	1/0.0	1	0/.0	ę -	243.8		1237.4	1
2 7	D 9759	T.	1200 42	ĩ	1545 8	1	749 7	:	11.77 7	•	510.0		1057 0	1
∔ T	D G230	ĭ -	1077.00	ä »	1707*0			ï.	17211	ĩ	J1017	ž *	102407	T
4 . T	8 2750	*	1900.82	7 1	1987.0		455.7	;	252. O	¥ 	707.2	*	1275.7	ı T
Ť		4	4700±02		1,0010		~~~~~		2		10/12	;	a an Fuir an F	Ţ
ĩ	B 8760	-	1901.70		2218.9		649.9		237.1	;	884.9	4	1331.9	Ţ
ĩ		÷				1				÷				Ŧ
Ī	B 8261	1	1903.50	ł	2458.5		617.2	:	291.0		908.2	:	1550.3	ī
Ĩ	-	1				3		ų		ł		ł		Ī
I	B 8417	1	1913	\$	1388.6	5	68.7		54.2	-	122.9	1	1265.7	I
I		;		4		:		1		5		à		I
I	B 8262	5	1926.50	5	1646.8	ŧ	249.1	:	112.2	:	361.3	ĩ	1285.5	T
I		:		1		2		ä		1		;		ĩ
Ţ	B 8418	2	1930.25	ł	674.5	à	193.8	\$	95,7	ź	289.5	ţ	385.0	I
I		:		;		1		2		;	•	ŧ		I
I	B 8257	:	1866-71	* R		:		:		5		ž		I
1	pil	1	DST-2	2		1		:		1		4		ľ
***		===		= = :	*******	***	*******	===	******	z # #	******	====	******	= =

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### COMPOSITION IN % OF MATERIAL EXTRACTED FROM THE ROCK

saa	= ≠ ≈	*******			= = =		; ± ;	*******	£ 3	******	# 3			******	= 1		=
I			;		:	Sat	:	Aro :	F.	HC	;	SAT :	1	Non HC	:	HC	I
I	I۲	(U-No	3	DEFTH	2		2				Ĭ		1		4		ĩ
1			1	(		EOM	1	EOM :	1	EOM	1	Aro :		EOM	1	Non HC	1
1 7			5	(m)	5						:	******			:		1. 
1						****			*			د بد بر رب مر مر مد ت د د			•		T T
I	₿	8414	1	1852.25	3	11.9	1	5.9	< ;	17.8	4	203.0		82.2	:	21.6	I
ž T	B	8415	7	1862.47	1	9.8	5 5	4.3	•	14.1	*	230.3	;	85.9	•	16.4	1
i I	B	8416	1	1881.71	1	11.7	1	4.5 1	•	16.2	1	260.9	i [	83.8	1	17,4	ı I
1 I 7	B	8258	:	1897.68	1	23.5	Ť	.9.1		32.6	T T	258.0		67.4	1	48.4	1 [ 7
i I T	₿	8259	;	1900.82	5	23.0		12.7	2	35.7	; ; ;	180.6	i ; ;	64.3	:	55.4	L I T
ı I T	₿	8260	5	1901.70	2	29.3	1	10.7 :		40.0	:	274.2	1	60.0	1	66.6	I
I T	B	8261	4 [	1903.50	:	25.1	5 7 7	11.8	*	36.9		212.1	•	63.1	2	58.6	Ī
I T	B	8417	*	1913	1	4.9	•	3.9		8.9	:	126.8	•	91.1	:	9.7	I T
Ī	B	8262	:	1926.50	:	15.1	1	6.8	*	21.9		222.0	;	78.1	•	28.1	Î
Ì	в	8418	¢ ;	1930.25	:	28.7	*	14.2		42.9	:	202.5	;	57.1	;	75.2	Ī
I	B	8257	; ; ;	1866-71	1	14.1	:	5.8 :	*	19.9	*	241.5	1	80.1	-	24.9	Į
z==:			₹ ≓⇒:	101-2 122222222	; :22		3 (3):		:	NGERIAG	3 ⊈a	***********		******	: # :	*******	

DATE: 7 - 1 - 85.

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- 25 -

TABLE 7



TABULATION OF DATA FROM THE GASCHROMATOGRAMS

222: 7	z = :		NEETU			OSSPENSE SDICTANC	***********	*******	==: T
r T	TÝ	al No.	DELLIN	ë e	FRISTANC.		CPT-1	CPI-2	4 T
I	**		(m)	;	n-C17	PHYTANE		w	Ĩ
Ia≈:	* 3: 5	******		*****		********		*******	:≡ <u>1</u> 7
I I T	B	8414	1852.25	4 2 2	0.8	0.9	1.1	1.3	I
I I	B	8415	1862.47		0.6	1.7	1.1	1.2	Î I
Î I	B	8416	1881.71	đ	0.6	1.7	1.1	1.2	II
I I	B	8258	1897.68	H H H	0.8	1.4	1.0	1.0	I I
ĭ I	B	8259	1900.82		0.8	1.5	1.0	1.0	I I
I I	B	8260	1901.70	1	0.8	1.4	0.9	1.0	I I
I I	B	8261	1903.50	5 2	0.8	1.4	1.0	1.0	I
I I	B	8417	1913	; ;	0.6	1.7	1.0	1.1	I I
I	B	8262	1926.50	:	0.6	1.7	1.1	1.1	1 1
I I	B	8418	1930.25	*	0.8	1.5	1.0	<b>i.i</b>	1 I
I I	B	8257 ail	1866-71 DST-2	\$ \$	0.7	2.4		-	I I

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IKU no.	Depth (m)	MNR	DMNR	MPI 1	MPI 2
P_9/1/	1852.25	1.2	2.5	0.83	0.83
D-0414 D-0414	1862-47	1.3	2.8	0.84	0.88
B-0415 R-8416	1881.71	1.1	3.1	0.87	0.88
B-8258	1899.68	1.4	3.6	0.92	0.94
B-8259	1900.82	1.5	3.4	0.93	0.93
B-8260	1901.70	1.4	3,2	1.02	1.04
B-8261	1903.50	1.4	3.0	0.95	0.95
B-8417	1913.00	1.2	2.4	0.85	0.85
B-8262	1926.50	1.2	3.0	0.90	0.95
B-8418	1930.25	1.6	3.3	0.81	0.81
B-8257	1866-71	1.6	3.8	-	-
oil					

## Table 8. Aromatic ratios.

Table 9. Molecular ratios calculated from terpane and sterane mass chromatograms. Maturity ratios, (1)2)1 3)1 4)1 1 -1 1 IKU No. I DEPTH Ιαβ/αβ+βα Ι %22S I % ββ Ι \$205 1 (m) 1 1 I + 1 ł \* 88415 1862.47 0.84 60.00 45.45 25.69 1903.50 0.91 61.11 74.00 88261 52.88 57,14 77.42 88257 62,50 0.91 oil

1) E/E+F in m/z 191.

2) % distribution between first and second

elution isomers of doublet J (m/z 191)

3) 2(r+s)/(q+t+2(r+s)) in m/z 217.

4) q/q+t in m/z 217.





- 29 -



### Table 10.

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=	Molecula and ster Source (	ar i rane chai	ratios e mass racteri	cal chr sti ===	culate omatog c and i	d ran na :	from te ns, turity	rpa rat	ine ios.		
ł		1			1	)	2	)	3)	4)!	5)1
ł	IKU No.	I.	DEPTH	1	Q/E	ł	Tm/Ts	I.	XZE I	a∕a+j l	Z/E I
I		1.	(m)	1		ł		I	I	1	I
#	88415 88261	<b>:</b> 2; = 3	1862.47 190 <u>3</u> .50	23,3	0.03	==	1.29 0.94	= = =	0.03 0.09	0.42	0.00
	B8257		oil		1.18		0.23		0.04	0.52	0.00

1) Relative abundance of tricyclic terpanes(Q/E in m/z 191).

2) B/A in m/z 191.

N. .

3) Relative abundance of unknown(X/E in m/z 191),
4) Relative abundance of C27 rearranged steranes(a/a+j),
5) Relative abundance of bisnorhopane(Z/E in m/z 191).



Table 11. Maturity ratios from aromatic steranes.

Sample	Depth (m)	m/z 253 <sup>%C</sup> 21 <sup>/C</sup> 21 <sup>+C</sup> 28,29	m/z 231 <sup>%C</sup> 20 <sup>/C</sup> 20 <sup>+C</sup> 26,27	%TRI/TRI+MONO
B-8257	011	-	-	-
B-8415	1862.47	25.0	-	61.1
B-8261	1903.50	30.8	23.6	70.2

## 142/p/jb1/22

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-	31	-

<u>Table 12.</u> $\delta^{13}$ C is	sotope data.			
Sample	Depth (m)	SAT	ARO	Extract/011
B-8214	1852.25	-30.2	-28.0	-29.3
B-8215	1862.47	-29.6	-28.6	-28.8
B-8216	1881.71	-29.6	-28.4	-29.2
B-8258	1899.68	-29.7	-28.3	-29.4
B-8259	1900.82	-29.9	-28.7	-29.0
B-8260	1901.70	-29.5	-28.4	-29.1
B-8261	1903.50	-29.7	-28.9	-29.2
B-8417	1913.00	-	-28.4	-29.2
B-8262	1926.50	-29,7	-	-29.1
B-8418	1930.50	-29.6	-28.3	-29.3
B-8257 (oil)	1866-71	-29.4	-27.7	-28.8
Mean values core	s	-29.7	-28.4	-29.2
Max. variation		0.7	0.9	0.6

## 142/p/jb1/23

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Figure 1

- 32 -

Whole oil GC

## 142/p/jb1/24

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Figure 2

- 34 -

C<sub>2</sub>-C<sub>8</sub> hydrocarbon GC

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## 142/p/jb1/25

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Organic carbon, Weight %

Figure 3. Quantity of organic carbon and extractable  $C_{15}$  + hydrocarbons in sediments (after Hunt, 1979)


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# Branched/cyclic hydrocarbon GC

142/p/jb1/27

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## Figure 6

### Aromatic hydrocarbon GC

### 142/p/jb1/28

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Aromatic hydrocarbons DMN B-8415 signal {%} :2.723 2.60-Core MP 1862.47m 2.40-TMN 2.20-Aromatic steranes imum Åå: Max 2.00-Ċ ò MMAANNAMM I. Inject 52 1.80-\$ 4 0 .. <del>เ</del> 1.60 -ВоХ MN Samp 7121/ 1-40-15A 2 CHRNNEL 9884 Plot 1-20-PLOT Scal Ž Analys Sample 1.00-DATA Full į 9 79 14 19 24 29 59 74 39 44 49 54 69 34 6.

on 05/Dec/84

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MN Aromatic hydrocarbons Injection £: 1 Maximum signal <%> :5.724 DMN B-8418 Core 50.00-1930.25m 40-00-**Aromatic steranes** MP TMN way was all was 1 30-00-60 a, ÷ £ ... 20-00 4 . Ход on 05/Dec/84 ₽ . :749B6418F Samp] me :B-8418,7121/7 2 10-00 CHRINNEL 12:21 Plot s Name Name a t PL0 Sca Rnalysi Sample reated 0.00-DATA Data 79 4 9 14 19 24 29 34 39 49 54 59 64 69 74 44

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F0/001/00

1 a 1 a 1

Figure 7a.

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Mass chromatograms representing terpanes (m/z 191)

Α	T <sub>s</sub> , 18α(H)-trisnorneohopane	$C_{27}H_{46}$	(111)
В	$T_m$ , 17 $\alpha$ (H)-trisnorhopane	C <sub>27</sub> H <sub>46</sub>	(I,R=H)
C	17a(H)-norhopane	C29H50	(I,R=C <sub>2</sub> H <sub>5</sub> )
D	17g(H)-normoretane	C <sub>29</sub> H <sub>50</sub>	(II,R=C <sub>2</sub> H <sub>5</sub> )
E	17α(H)-hopane	C <sub>30</sub> H <sub>52</sub>	(I,R=C <sub>3</sub> H <sub>7</sub> )
F	17β(H)-moretane	C <sub>30</sub> H <sub>52</sub>	(II,R=C <sub>3</sub> H <sub>7</sub> )
G	17α(H)-homohopane (22S)	C <sub>31</sub> H <sub>54</sub>	(I,R=C4Hg)
H	17α(H)-homohopane (22R)	C <sub>31</sub> H <sub>54</sub>	(I,R≐C <sub>4</sub> H <sub>9</sub> )
	+ unknown triterpane (gammacerane?)	01 04	
I	17β(H)-homomoretane	C31H54	(II,R=C <sub>4</sub> H <sub>9</sub> )
J	$17\alpha(H)$ -bishomohopane (22S,22R)	С <sub>32</sub> Н <sub>56</sub>	$(1, R=C_5H_{11})$
К	$17\alpha(H)$ -trishomohopane (22S,22R)	C <sub>33</sub> H <sub>58</sub>	$(1, R=C_6H_{13})$
L	$17\alpha(H)$ -tetrakishomohopane (22S,22R)	$C_{34}H_{60}$	$(I,R,=C_7H_{15})$
М	$17\alpha(H)$ -pentakishomohopane (22S,22R)	C <sub>35</sub> H <sub>62</sub>	(I,R=C <sub>8</sub> H <sub>17</sub> )
Z	bisnorhopane	C <sub>28</sub> H <sub>48</sub>	•
X	unknown triterpane	C <sub>30</sub> H <sub>52</sub>	
P	tricyclic terpane	C <sub>23</sub> H <sub>42</sub>	$(IV,R=C_4H_q)$
Q	tricyclic terpane	C <sub>24</sub> H <sub>44</sub>	$(IV, R=C_5H_{11})$
R	tricyclic terpane (17R,17S)	C <sub>25</sub> H <sub>46</sub>	$(IV, R=C_6H_{13})$
S	tetracyclic terpane	$C_{24}H_{42}$	(V)
Т	tricyclic terpane (17R,17S)	C <sub>26</sub> H <sub>48</sub>	$(IV, R=C_7H_{15})$

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Figure 7b.

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Mass chromatograms representing steranes (m/z 21/ and	d 218)
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a	13β(H),17α(H)-diasterane (20S)	С <sub>27</sub> н <sub>48</sub>	(III,R=H)
b	13β(H),17α(H)-diasterane (20R)	C <sub>27</sub> H <sub>48</sub>	(III,R=H)
с	13a(H),17g(H)-diasterane (20S)	C <sub>27</sub> H <sub>48</sub>	(IV,R=H)
d	13μ(H),17β(H)-diasterane (20R)	C27H44	(IV,R=H)
e	136(H),17a(H)-diasterane (20S)	C20H50	(III,R≈CH <sub>2</sub> )
f	13β(H),17α(H)-diasterane (20R)	C28 50	(111,R=CH <sub>2</sub> )
9	13u(H),17β(H)-diasterane (20S)		'(IV,R=CH <sub>2</sub> )
	+ $14\alpha(H), 17\alpha(H)$ -sterane (20S)	28 30 CarHau	(1,R=H)
h	$13B(H), 17\alpha(H)$ -diasterane (20S)	C20HE2	(III,R=C <sub>2</sub> H <sub>c</sub> )
	+ 148(H),178(H)-sterane (20R)	23 52 C <sub>27</sub> Η <sub>40</sub>	(II,R=H)
i	14B(H), 17B(H)-sterane (20S)	C27 40 C27H40	(II,R=H)
	+ $13\alpha(H)$ , $17\beta(H)$ -diasterane (20R)		(IV,R=CH <sub>2</sub> )
j	$14\alpha(H), 17\alpha(H)$ -sterane (20R)	C <sub>27</sub> H <sub>40</sub>	(I,R=H)
k	$13B(H), 17\alpha(H)$ -diasterane (20R)	$C_{20}H_{E2}$	$(III, R=C_2H_E)$
1	13u(H), 17B(H)-diasterane (20S)		(III,R=C <sub>o</sub> H <sub>c</sub> )
m	14a(H), 17a(H)-sterane (20S)		(1,R=CH <sub>2</sub> )
n	13u(H), 17B(H)-diasterane (20R)	28 50 ConHere	(III,R=C,H <sub>c</sub> )
	+ 14β (H),17β (H)-sterane (20R)	29 52 C <sub>20</sub> H <sub>60</sub>	(11,R=CH <sub>2</sub> )
0	$14\beta(H), 17\beta(H)$ -sterane (20S)	28 50 C20HEO	(II,R=CH <sub>2</sub> )
p	$14u(H), 17\alpha(H)$ -sterane (20R)	28 50 دینطری	(1,R=CH <sub>2</sub> )
Ģ	$14\alpha(H), 17\alpha(H)$ -sterane (20S)	ເລັ້ນ ເ	(1,R≏C <sub>2</sub> H <sub>c</sub> )
r	$14\beta(H), 17\beta(H)$ -sterane (20R)	29 52 C.,, tig.,	(II,R=C <sub>2</sub> H <sub>r</sub> )
	+ unknown sterane	<i>4</i> 3 34	25
s	$14\beta(H), 17\beta(H)$ -sterane (20S)	C.,	(II,R=C <sub>o</sub> H <sub>c</sub> )
t	$14\beta(H), 17\beta(H)$ -sterane (20R)	29 52 C.,,,He.,	(I,R=C.,H <sub>c</sub> )
u	5u(H)-sterane	29 52 C.,H32	(V,R=C <sub>0</sub> H <sub>E</sub> )
v	5u(H)-sterane	مد 21 21 مە	(IV,R=C_H_)
		22 30	5.1









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# Figure 8

Mass chromatograms of aromatic hydrocarbons

TIC		-	total	ion chromatograms
m/z	92,106	-	a1ky1	benzenes
m∕z	142,156,170	-	alkyl	naphthalenes
m/z	178,192,206	-	alkyl	phenanthrenes
m/z	184,198,212	-	alkyl	dibenzothiophenes
m/z	231	-	triard	omatic steranes
m/z	253	-	monoai	romatic steranes

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Core samples

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Oil sample

142/n/ib1/30

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IHP K: B8257AR2 #600-1400 K1:184 m/z 184 109, B 8257 Qil 88. K Т ī IHP L: 882578R2 #680-1489 L1:198 100, m/z 198 B 8257 Oil L 



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