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	REPORT ON STABLE ISOTOPES On natural gases from wei		REV. NO.		
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	CLIENT REF. Ellen Sofie Moe	NUMBER OF ISSUES 15			
SUMMARY			DISTRIBUTION		
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PREPARED	BY Bjørg Andresen Einar M. Brevik Arne Råheim	1985-05-30 1985-05-30 1985-05-30	Bjørg Andreau Rina Suirle. Am Rani		
REVIEWED	ВҮ		1		
APPROVED	BY Karen Garder	1985-05-30	Karen Tarder		

1. ANALYTICAL PROCEDURE

The natural gases have been separated into the different gas components by a Carlo-Erba 4200 instrument. This gas chromatograph is equipped with a special injection loop in order to concentrate the samples, in the case of low concentration of the gas components. The hydrocarbon gas components were oxidized in separate Cu0-ovens in order to prevent cross contamination. The combustion products CO₂ and H_2O were frozen into collection vessels and separated.

The water was reduced with zinc metal in a sealed tube to prepare hydrogen for isotopic analysis. The isotopic measurements were performed on a Finnigan Mat 251 mass spectrometer. Our δ^{13} C value on NBS-22 is -29.77 <u>+</u> .06 o/oo.

2. RESULTS

The composition of the samples are given in Table 1. The results have not been normalized to 100%. The rest is air. The stable isotope results are given in Table 2.

Our uncertainty on the δ^{13} C value is estimated to be \pm 0.3 o/oo and includes all the different analysis step. The uncertainty on the δ D value is likewise estimated to be \pm 5 o/oo.

Sample	·	°1	°2	°3 1	i-C ₄	n-C ₄	co2
34/7-4	DST#1	80.0	8.8	5.0	0.45	1.20	0.42
34/7-3 A 14885	DST#2B	82.3	7.3	3.5	0.22	0.57	0.24
34/7-3 A 14667	DST#1	78.4	8.8	5.1	0.37	1.00	0.37

Table 1 Composition of Gas Samples from Well 34/7-4 and 34/7-3

<u>Table 2</u> Isotopic Composition of Gas Samples from Well 34/7-4 and 34/7-3

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	c ₁		с ₂	c ³	i-C ₄	n-C4	coz	
Sample	δ ¹³ C	δD	δ ¹³ c δ ¹⁸					
34/7-4 DST#1	~49.8	-147	-34.4	-31.6	-30.5	-31.9	-13.8 -13.	
34/7-3 DST#28 A 14885	-51.4	-203	-35.7	-32.8	-29.0	-32.1	-21.7 -23. <u>+</u> 0.5	
34/7-3 DST#1 A 14667	-51.9	-175	-36.6	-33.7	-29.3	-33.0	-15.4 -13.1	

3. INTERPRETATION

The δ^{13} C values of methane, ethane, propane and n-butane have been plotted on the maturation diagram by James (1983)^{*}, Figure 1. The three samples indicate a source LOM of about 11, e.g. that the gas was formed at a relatively high maturity in the oil window.

The carbon and hydrogen isotopic composition have been plotted in a δ^{13} C methane vs. δ D methane cross plot (Schoell 1983)^{**}, Figure 2. This indicates that the gases may have a biogenic component.

4. CONCLUSION

The carbon isotopic distribution between the hydrocarbon gas components methane, ethane, propane and n-butane indicates that the gas was formed at a relatively high maturity in the oil window.

The carbon and hydrogen isotopic composition of methane suggest that methane has a biogenic component.

- * James, Alan T. (1983): Correlation of Natural Gas by Use of Carbon Isotopic Distribution between Hydrocarbon Components, AAPG, Vol. 67, No. 7, July 1983.
- ** Schoell, M. (1983): Genetic Characterization of Natural Gases, AAPG, December 1983.

Enclosed is a copy of the isotopic composition of two gas samples from well 34/7-3 received from IKU and analysed at IFE March 1985.



KJELLER

IKU SINTEF-GRUPPEN Seksjon for Organisk Geokjemi Boks 1883 Jarlesletta 7001 TRONDHEIM

Attn. Torun Vinge

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Hovedkonto	r og forskningssenter
Boks 40, 21	007 Kjeller
Telefon:	(02)712560-713560
Teleks:	74 573 energ n
Telefax:	(02)71 55 53
Telegram:	Isotop Oslo
Bankgiro:	5102.05.00070
Postgiro:	5 33 96 01
Dato:	1985-03-29
Vår ref:	BjA/EBJ
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ISOTOPANALYSE AV 4 GASSPRØVER

Ved ankomst IFE var prøve B-9555 (Saga 34/7-3) knust og gass-sylinder B-9565 (Saga 34/7-4) tom.

Gass-sylinder B-9147 (Saga 34/7-3) inneholdt også lite gass, kun tilstrekkelig til en analyse. Isotopverdiene fra denne gassen viser en unormal sammensetning. Muligens kan dette være fordi sylinderen inneholdt lite gass eller fordi sylinderen inneholdt annen gass ved påfylling av prøvegassen.

På grunn av for lite prøve har vi heller ikke vært i stand til å analysere D/H for metan.

For å utføre en skikkelig analyse av prøvene må vi få tilsendt mer gass.

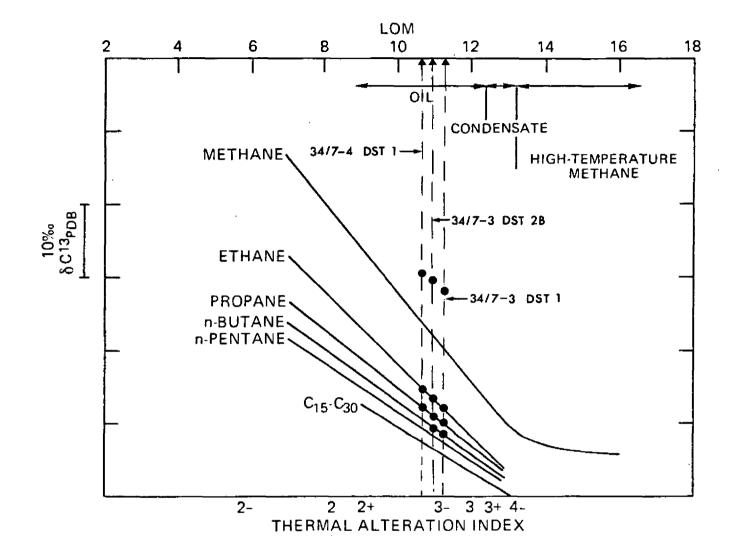
Analyseresultater:

Prøve	δ ¹³ C _{PDB}						
	с ₁	с ₂	с ₃	iC4	nC4		
B-9557 (Saga 34/7-3 DST 2}	-41.6	-31.9	-30.0	- 22 . 1	-29.4		
B-9147 (Saga 34/7-3 separator gass fra singel flash)	-51.2	-36.3	-34.0	-25.9	-34.0		

Med hilsen

Bjørg Anciresin. Bjørg Andresen





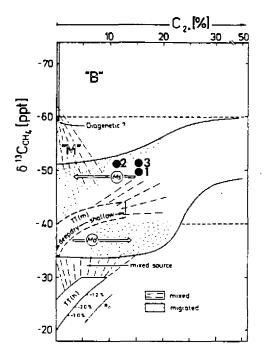
<u>Figure 1</u>. Carbon isotopic separations of the gas from well 34/7-4 and 34/7-3 are plotted on the maturity diagram (after James, 1983). A source LOM of about 11 is indicated for the gases.

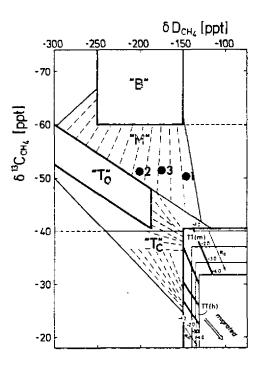
The calculated carbon isotopic separations between gas components are plotted on the vertical axis using a sliding scale that is simply the algebraic difference, in parts per mil, between the isotopic compositions of the natural gas components. The scale does not possess a fixed origin, but is oriented with the more depleted δ^{13} C values at the upper end. Use of this sliding scale allows the maturity of a gas to be assessed without prior knowledge of the isotopic composition of the gas source.



1. 34/ 7-4 DST 1 2. 34/ 7-3 DST 2B

3. 34/7-3 DST1





<u>Figure 2a</u>

Variations of molecular composition in natural gases related to the isotope variations of methane. Figure 2b

Carbon and hydrogen isotope variations in methanes.

The principle for the genetic characterization of natural gases is that the primary gases (8-biogenic gas, T-associated gas, TT-non-associated gas) are defined by fields of compositional variations. These primary gases may become mixed and form various mixtures "M" of intermediate composition. "TT(m)" and "TT(h)" are non-associated gases from marine source rocks and coal gases from N.W. Germany, respectively, compositional shifts due to migration are indicated by arrows Md (deep migration) and Ms (shallow migration), respectively. "T₀" are gases associated with petroleum in an initial phase of formation. "T₀" are gases associated with condensates. (Schoell 1983).